

Biosynthesis of the Antitumour Catharanthus Alkaloids: The Fate of the 21' α -Hydrogen of Anhydrovinblastine

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[21' α - ^3H ,methyl- ^{14}C]Anhydrovinblastine is incorporated into vinblastine by cell-free preparations of *Catharanthus roseus* without loss of ^3H .

The chemotherapeutic importance of the bis-indole dimeric alkaloids, vinblastine (**1**) and vincristine (**2**), from *C. roseus* has given impetus to both synthetic^{1,2} and biosynthetic³⁻⁸ studies of these and the related alkaloids, leurosine (**3**) and leurosine (**4**). Incorporation experiments with whole plants^{3,4} and cell-free systems^{5b} have established the role of vindoline (**8**) and catharanthine (**10**) as precursors of the *Aspidosperma* and *Iboga* segments of the dimers, respectively. Anhydrovin-

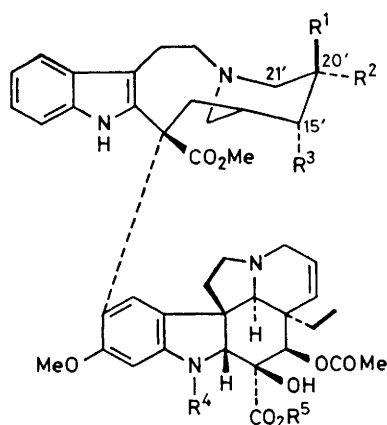
blastine (**5**), radiolabelled in either, but not both, of the *Aspidosperma* or *Iboga* derived segments, has been shown to be incorporated into (**1**) by cell-free preparations.^{5,6a} In addition, singly-labelled 20'-deoxyleurosine (**6**) has been incorporated into (**1**) by intact *C. roseus* plants.⁷

While these results suggest a biosynthetic route (**8**) + (**10**) \rightarrow (**5**) \rightarrow (**6**) \rightarrow (**1**) [path (a) in Scheme 1] similar to that previously proposed by Potier² on the basis of synthetic analogy,

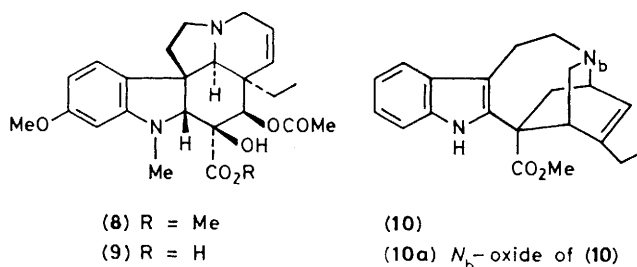
Table 1. Incorporation of anhydrovinblastine (5) into vinblastine (1) by cell-free extracts^a of *C. roseus*.

Expt	Precursor	Anhydrovinblastine fed (d.p.m.)		³ H/ ¹⁴ C ratio	(1) isolated (d.p.m.) ^e		³ H/ ¹⁴ C ratio	% incorp. ^f
		³ H	¹⁴ C		³ H	¹⁴ C		
1	[21'- ³ H, methyl- ¹⁴ C](5) ^e	5.03 × 10 ⁶	7.37 × 10 ⁵	6.83 ± 0.05	8.59 × 10 ⁴	1.23 × 10 ⁴	6.98 ± 0.24	1.67
2	[21'- ³ H, methyl- ¹⁴ C](5) ^e	4.70 × 10 ⁶	5.70 × 10 ⁵	8.25 ± 0.05	3.69 × 10 ⁴	4.56 × 10 ³	8.10 ± 0.30	0.8
3 ^b	[21'- ³ H] ^a (5)	8.24 × 10 ⁷	—	—	1.32 × 10 ⁴	—	—	0.02

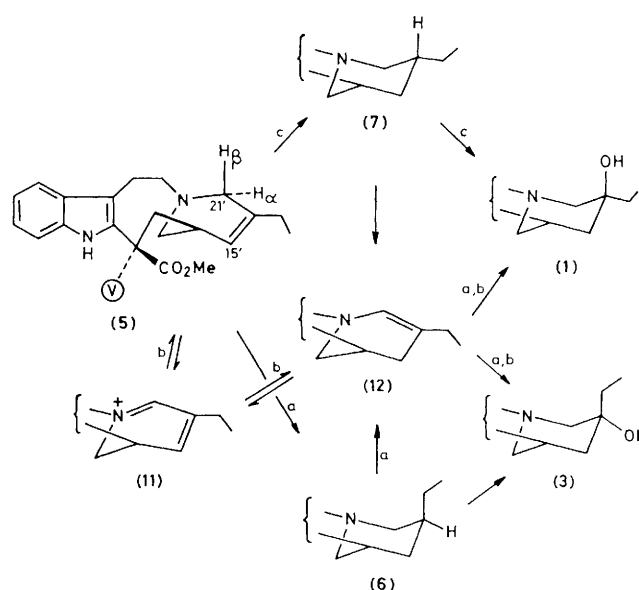
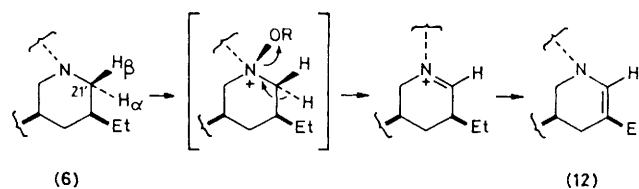
^a Cell-free extracts were prepared as described (A. I. Scott and S-L. Lee, *J. Am. Chem. Soc.*, 1975, **97**, 6906). Approximately 10 g fresh leaves/10 ml of 0.05 M tris-maleate buffer (pH 7.0) were used in expts 1 and 3. The concentration of plant material was halved in expt 2. ^b Extract boiled for 5 min prior to feeding. ^c Specific activities 4.01 mCi mmol⁻¹ for ³H; 0.22 mCi mmol⁻¹ for ¹⁴C. ^d Specific activity 38.25 mCi mmol⁻¹. ^e At the end of each expt vinblastine sulphate (8–10 mg) was added, the extract was adjusted to pH 11 (NH₄OH) and extracted (CH₂Cl₂); (5) and (1) were separated by t.l.c., (1) was further purified by h.p.l.c. and crystallised to constant activity as its 0.5 MeOH:0.5 Et₂O solvate. ^f No adjustment was made for recovered precursor.



- (1) R¹ = OH, R² = Et, R³ = H, R⁴ = R⁵ = Me
 (2) R¹ = OH, R² = Et, R³ = H, R⁴ = CHO, R⁵ = Me
 (3) R¹ = Et, R² = OH, R³ = H, R⁴ = R⁵ = Me
 (4) R¹ = Et, R², R³ = -O-, R⁴ = R⁵ = Me
 (5) R¹ = Et, Δ^{15'(20')}, R⁴ = R⁵ = Me
 (6) R¹ = Et, R² = H, R³ = H, R⁴ = R⁵ = Me
 (7) R¹ = H, R² = Et, R³ = H, R⁴ = R⁵ = Me



Guéritte⁷ has argued that they are equally compatible with the existence of a biogenetic grid in which the conjugated iminium salt (11), derived from a Polonovski type reaction of (8) and the *N*-oxide (10a) is in equilibrium with the 1,2-reduction product anhydrovinblastine (5) and the 1,4-reduction product (12) [path (b)]. Hydration of the enamine (12) could then give rise to (1) and (3). A feature common to both of the suggested pathways from (5) to (1) is the loss of one of the 21'-hydrogens of (5) by a process which could be mediated by the biological equivalent of Polonovski elimination of the corresponding *N*-oxide, a possibility advanced earlier.² If this were the case then loss of the 21' α -H should be expected, as only this hydrogen can adopt an antiperiplanar orientation relative to the oxygen of (6) [or (5)] *N*-oxide (Scheme 2). To test the possible intervention of an *N*-oxide (albeit indirectly) it was required to prepare anhydrovinblastine stereospecifically labelled with ³H in the 21' α -position.

**Scheme 1****Scheme 2**

Inspection of models of the coupling reaction product (11)¹ indicated that the steric congestion of the β -face of the dihydropyridinium ring might force hydride reduction to occur predominantly from the less hindered α -face. Gratifyingly, treatment of (11) with NaB[²H]₄ in methanol afforded mono-deuterio-(5), the ¹H n.m.r. spectrum (360 MHz) of which showed absence of the 21' α -H doublet at δ 3.52 and the collapse of the 21' β -H doublet at δ 3.27 (*J* 16 Hz) to a singlet. [21' α -³H]Anhydrovinblastine (5) was prepared in a similar manner using NaB[³H]₄.⁵

Administration of [21' α -³H, methyl-¹⁴C](5) (R⁵ = ¹⁴CH₃),[†] to cell-free extracts of mature *C. roseus* leaves followed by

[†] [methyl-¹⁴C](5) (R⁵ = ¹⁴CH₃) was prepared by treatment of vindolic acid (9) with [¹⁴C]diazomethane, coupling¹ of the resultant [methyl-¹⁴C](8) (R = ¹⁴CH₃) with (10a) and reduction of the product with NaBH₄.

isolation, afforded (1) with no significant change in $^3\text{H}/^{14}\text{C}$ ratio (expts 1 and 2, Table 1), showing that the $21'\alpha\text{-}^3\text{H}$ was retained in the transformation.

These results provide the first unambiguous demonstration of intact incorporation of anhydrovinblastine into (1). Retention of the tritium at $21'\alpha$ indicates that the transformation (6) \rightarrow (12) or (5) \rightarrow (11) by *trans*-elimination involving the corresponding *N*-oxide (Scheme 2) are unlikely steps in the pathway.[‡] While this does not preclude the possibility that *cis*-elimination might be involved by a different mechanism of dehydrogenation, we suggest that the sequences (5) \rightarrow (6) \rightarrow (12) \rightarrow (1) [path (a)] and (5) \rightarrow (11) \rightarrow (12) \rightarrow (1) [path (b)] mediated by the *N*-oxide route shown in Scheme 2 do not appear likely as the major pathways from (5) to (1) *in vivo*.

A number of possible pathways compatible with the above results still remain: (a) direct hydration of the $\Delta^{15'(20')}$ double bond of (5), (b) reduction of (5) to 20'-deoxyvinblastine (7) and hydroxylation with *retention* of configuration [path (c) in Scheme 1], and (c) reduction to 20'-deoxyeuosidine (6) followed by hydroxylation with *inversion* of configuration.⁹

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[‡] Complete loss of ^3H might be expected for such a mechanism only if loss of the $21'\alpha\text{-H}$ were not a rate determining step. For the antithetical case an abnormal ^3H isotope effect of $>25:1$ would be required to invalidate this result (ref. 10).